

2-[(1*E*,4*E*)-2-(4-Methoxyphenyl)-3*H*-benzo[*b*][1,4]diazepin-4-yl]phenol

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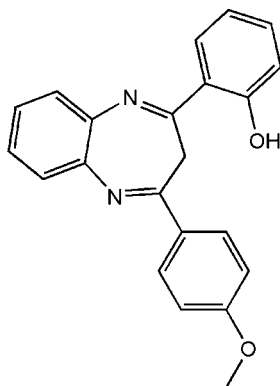
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.116; data-to-parameter ratio = 16.4.

The title compound, $\text{C}_{22}\text{H}_{18}\text{N}_2\text{O}_2$, is a biologically active 1,5-benzodiazepine derivative, containing three planar six-membered rings and one nonplanar seven-membered ring. The six-membered rings *A* (phenol), *B* (fused benzene) and *C* (methoxyphenyl) are oriented with respect to each other at dihedral angles of 39.22 (2) (*A/B*), 87.31 (3) (*A/C*) and 54.42 (3)° (*B/C*). The seven-membered ring adopts a near-boat conformation. In the crystal structure, intramolecular $\text{O}-\text{H}\cdots\text{N}$ and intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds are present.

Related literature

For bond-length data, see: Allen (2002); Bruno *et al.* (2004). For general background, see: Krapcho & Turk (1966); Gringauz (1999); Butcher & Hamor (1985); Armarego (1977). For related literature, see: Ahmed *et al.* (1990).



Experimental

Crystal data

$\text{C}_{22}\text{H}_{18}\text{N}_2\text{O}_2$
 $M_r = 342.38$
 Monoclinic, $P2_1/n$
 $a = 9.8203$ (7) Å
 $b = 16.2920$ (12) Å
 $c = 11.0178$ (8) Å
 $\beta = 105.203$ (1)°
 $V = 1701.1$ (2) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 298$ (2) K
 $0.33 \times 0.30 \times 0.27$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.897$, $T_{\max} = 0.980$
 9963 measured reflections
 4220 independent reflections
 3022 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.017$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.116$
 $S = 1.01$
 4220 reflections
 239 parameters
 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.20$ e Å⁻³
 $\Delta\rho_{\min} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| <i>D</i> -H \cdots <i>A</i> | <i>D</i> -H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> -H \cdots <i>A</i> |
|-----------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| O1-H1 \cdots N2 | 0.92 (2) | 1.71 (2) | 2.5563 (17) | 151.5 (19) |
| C14-H14A \cdots O2 ⁱ | 0.93 | 2.53 | 3.406 (2) | 156 |

Symmetry code: (i) $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HK2293).

References

- Ahmed, R., Malik, M. A. & Zia-ul-Haq, M. (1990). *J. Chem. Soc. Pak.* **12**, 352–354.
 Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Armarego, W. L. F. (1977). *General Heterocyclic Chemistry*, edited by E. C. Taylor & A. Weissberger. New York: John Wiley and Sons.
 Bruker (1998). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (1999). *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruno, I. J., Cole, J. C., Kessler, M., Luo, J., Motherwell, W. D. S., Purkis, L. H., Smith, B. R., Taylor, R., Cooper, R. I., Harris, S. E. & Orpen, A. G. (2004). *J. Chem. Inf. Comput. Sci.* **44**, 2133–2144.
 Butcher, H. J. & Hamor, T. A. (1985). *Acta Cryst.* **C41**, 1081–1083.
 Gringauz, A. (1999). *Introduction to Medicinal Chemistry* pp. 578–579. New York: Wiley-VCH.
 Krapcho, J. & Turk, C. F. (1966). *J. Med. Chem.* **9**, 191–195.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

supplementary materials

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2-[(1*E*,4*E*)-2-(4-Methoxyphenyl)-3*H*-benzo[*b*][1,4]diazepin-4-yl]phenol

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Comment

The title compound belongs to an important class of the pharmacologically pre-eminent 1,5-benzodiazepines which are extensively studied for medicinal activities (Krapcho & Turk, 1966). In recent years, benzodiazepines have replaced barbiturates which were used once for the purpose of hypnotic effects, owing to their less toxic and less severe withdrawal effects (Gringauz, 1999). The importance of 1,5-benzodiazepines, in particular is evident from the pharmaceutical application of Globazam (Butcher & Hamor, 1985). The difficulties encountered in the cyclization of these seven-membered heterocycles limited their structural studies. In view of the importance of this class of compounds, the title compound, (I), has been synthesized and its crystal structure is reported here.

A perspective view of (I) is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.28, November 2006; Allen, 2002; Mogul, Version 1.1; Bruno *et al.*, 2004). The seven membered diazepine ring has a fragment N1—C10—C11—N2, which is conjugated with the adjacent benzene ring. Like azepines, diazepines are not planar and generally adopt boat conformation (Armarego, 1977).

The rings A (C1—C6), B (C10—C15) and C (C16—C21) are, of course, planar and dihedral angles between them are A/B = 39.22 (2)°, A/C = 87.31 (3)° and B/C = 54.42 (3)°. The seven membered ring D (N1/N2/C7—C11) is non-planar and adopts nearly boat conformation.

In the crystal structure, intramolecular O—H···N and intermolecular C—H···O hydrogen bonds (Table 1) link the molecules (Fig. 2), in which they may be effective in the stabilization of the structure.

Experimental

1-(2-hydroxyphenyl)-3-(4-methoxyphenyl)propane-1,3-dione (2.7 g, 10 mmol), prepared according to the reported procedure (Ahmed *et al.*, 1990), was subjected to *cyclo*-condensation with phenylene diamine (1.08 g, 10 mmol) using toluene (100 ml) as solvent to get the title compound (yield; 62%, m.p.481–483 K). Yellow single crystals suitable for X-ray analysis were obtained by slow evaporation of dichloromethane solution.

Refinement

H1 (for OH) was located in difference syntheses and refined isotropically [O1—H1 = 0.92 (2) Å and $U_{\text{iso}}(\text{H}) = 0.088$ (6) Å²]. The remaining H atoms were positioned geometrically, with C—H = 0.93, 0.97 and 0.96 Å for aromatic, methylene and methyl H atoms, and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$, where $x = 1.5$ for methyl H, and $x = 1.2$ for all other H atoms.

Figures

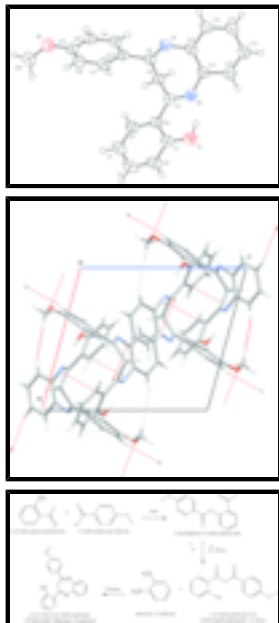


Fig. 1. The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Fig. 2. A packing diagram for (I). Hydrogen bonds are shown as dashed lines.

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Crystal data

$C_{22}H_{18}N_2O_2$

$M_r = 342.38$

Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

$a = 9.8203$ (7) Å

$b = 16.2920$ (12) Å

$c = 11.0178$ (8) Å

$\beta = 105.203$ (1)°

$V = 1701.1$ (2) Å³

$Z = 4$

$F_{000} = 720$

$D_x = 1.337$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 3874 reflections

$\theta = 2.3$ – 28.3 °

$\mu = 0.09$ mm⁻¹

$T = 298$ (2) K

Block, yellow

$0.33 \times 0.30 \times 0.27$ mm

Data collection

Bruker SMART CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293$ (2) K

ω and φ scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

$T_{\min} = 0.897$, $T_{\max} = 0.980$

4220 independent reflections

3022 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.017$

$\theta_{\text{max}} = 28.3$ °

$\theta_{\text{min}} = 2.3$ °

$h = -12 \rightarrow 12$

$k = -21 \rightarrow 20$

9963 measured reflections

$l = -14 \rightarrow 7$

Refinement

Refinement on F^2

Secondary atom site location: difference Fourier map

Least-squares matrix: full

Hydrogen site location: inferred from neighbouring sites

$R[F^2 > 2\sigma(F^2)] = 0.040$

H atoms treated by a mixture of independent and constrained refinement

$wR(F^2) = 0.116$

$$w = 1/[\sigma^2(F_o^2) + (0.0582P)^2 + 0.3204P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$S = 1.01$

$(\Delta/\sigma)_{\max} = 0.001$

3928 reflections

$\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$

239 parameters

$\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

Primary atom site location: structure-invariant direct methods

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|---------------|-------------|--------------|----------------------------------|
| O1 | -0.03533 (12) | 0.86806 (6) | 0.61876 (11) | 0.0555 (3) |
| H1 | 0.039 (2) | 0.8542 (13) | 0.5869 (19) | 0.088 (6)* |
| O2 | 0.72160 (11) | 1.13972 (6) | 1.11673 (10) | 0.0535 (3) |
| N1 | 0.51345 (11) | 0.89159 (7) | 0.66528 (10) | 0.0416 (3) |
| N2 | 0.19812 (11) | 0.87562 (7) | 0.55763 (11) | 0.0411 (3) |
| C1 | -0.00688 (14) | 0.94474 (8) | 0.66394 (13) | 0.0415 (3) |
| C2 | -0.10065 (16) | 0.98059 (9) | 0.72371 (15) | 0.0519 (4) |
| H2A | -0.1803 | 0.9517 | 0.7298 | 0.062* |
| C3 | -0.07683 (17) | 1.05816 (9) | 0.77385 (15) | 0.0560 (4) |
| H3A | -0.1393 | 1.0808 | 0.8150 | 0.067* |
| C4 | 0.03994 (17) | 1.10289 (9) | 0.76347 (14) | 0.0516 (4) |
| H4A | 0.0564 | 1.1552 | 0.7979 | 0.062* |
| C5 | 0.13119 (15) | 1.06868 (8) | 0.70144 (13) | 0.0433 (3) |
| H5A | 0.2078 | 1.0995 | 0.6926 | 0.052* |
| C6 | 0.11271 (13) | 0.98942 (8) | 0.65144 (11) | 0.0365 (3) |
| C7 | 0.21520 (12) | 0.95158 (8) | 0.59207 (12) | 0.0364 (3) |

supplementary materials

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|------|--------------|--------------|--------------|------------|
| C8 | 0.34185 (13) | 0.99771 (8) | 0.57406 (12) | 0.0383 (3) |
| H8A | 0.3310 | 1.0562 | 0.5848 | 0.046* |
| H8B | 0.3540 | 0.9879 | 0.4907 | 0.046* |
| C9 | 0.46674 (13) | 0.96498 (8) | 0.67378 (12) | 0.0375 (3) |
| C10 | 0.44695 (14) | 0.83856 (8) | 0.56771 (12) | 0.0401 (3) |
| C11 | 0.29969 (14) | 0.83181 (8) | 0.51475 (13) | 0.0409 (3) |
| C12 | 0.24887 (16) | 0.77175 (9) | 0.42316 (15) | 0.0513 (4) |
| H12A | 0.1520 | 0.7662 | 0.3899 | 0.062* |
| C13 | 0.33820 (17) | 0.72090 (9) | 0.38116 (16) | 0.0557 (4) |
| H13A | 0.3023 | 0.6837 | 0.3167 | 0.067* |
| C14 | 0.48251 (17) | 0.72541 (9) | 0.43547 (16) | 0.0548 (4) |
| H14A | 0.5434 | 0.6901 | 0.4091 | 0.066* |
| C15 | 0.53528 (15) | 0.78209 (9) | 0.52820 (14) | 0.0488 (3) |
| H15A | 0.6318 | 0.7832 | 0.5659 | 0.059* |
| C16 | 0.53499 (13) | 1.01454 (8) | 0.78517 (12) | 0.0375 (3) |
| C17 | 0.51971 (15) | 1.09936 (8) | 0.79032 (14) | 0.0457 (3) |
| H17A | 0.4673 | 1.1269 | 0.7195 | 0.055* |
| C18 | 0.58070 (15) | 1.14386 (8) | 0.89838 (14) | 0.0478 (3) |
| H18A | 0.5683 | 1.2004 | 0.9000 | 0.057* |
| C19 | 0.65993 (13) | 1.10332 (8) | 1.00349 (12) | 0.0411 (3) |
| C20 | 0.68271 (15) | 1.01928 (8) | 0.99855 (13) | 0.0451 (3) |
| H20A | 0.7404 | 0.9925 | 1.0677 | 0.054* |
| C21 | 0.62032 (13) | 0.97595 (8) | 0.89211 (12) | 0.0409 (3) |
| H21A | 0.6348 | 0.9196 | 0.8906 | 0.049* |
| C22 | 0.68065 (19) | 1.22061 (10) | 1.13736 (17) | 0.0611 (4) |
| H22A | 0.7318 | 1.2384 | 1.2197 | 0.092* |
| H22B | 0.5813 | 1.2218 | 1.1310 | 0.092* |
| H22C | 0.7010 | 1.2566 | 1.0753 | 0.092* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|------------|------------|-------------|-------------|------------|-------------|
| O1 | 0.0516 (6) | 0.0463 (6) | 0.0779 (8) | -0.0086 (5) | 0.0334 (6) | -0.0049 (5) |
| O2 | 0.0569 (6) | 0.0490 (6) | 0.0512 (6) | 0.0015 (5) | 0.0083 (5) | -0.0068 (5) |
| N1 | 0.0382 (6) | 0.0435 (6) | 0.0450 (6) | 0.0038 (5) | 0.0145 (5) | -0.0012 (5) |
| N2 | 0.0369 (6) | 0.0407 (6) | 0.0487 (6) | 0.0002 (4) | 0.0167 (5) | -0.0025 (5) |
| C1 | 0.0419 (7) | 0.0408 (7) | 0.0450 (7) | 0.0035 (5) | 0.0169 (6) | 0.0073 (6) |
| C2 | 0.0490 (8) | 0.0510 (8) | 0.0648 (9) | 0.0061 (6) | 0.0313 (7) | 0.0131 (7) |
| C3 | 0.0616 (9) | 0.0541 (9) | 0.0633 (10) | 0.0170 (7) | 0.0360 (8) | 0.0089 (7) |
| C4 | 0.0619 (9) | 0.0415 (7) | 0.0558 (9) | 0.0115 (6) | 0.0233 (7) | 0.0014 (6) |
| C5 | 0.0430 (7) | 0.0392 (7) | 0.0494 (8) | 0.0038 (5) | 0.0155 (6) | 0.0042 (6) |
| C6 | 0.0351 (6) | 0.0379 (6) | 0.0377 (6) | 0.0056 (5) | 0.0119 (5) | 0.0059 (5) |
| C7 | 0.0339 (6) | 0.0393 (6) | 0.0367 (6) | 0.0026 (5) | 0.0105 (5) | 0.0040 (5) |
| C8 | 0.0379 (6) | 0.0390 (6) | 0.0401 (7) | 0.0000 (5) | 0.0143 (5) | 0.0037 (5) |
| C9 | 0.0329 (6) | 0.0404 (7) | 0.0426 (7) | -0.0004 (5) | 0.0159 (5) | 0.0038 (5) |
| C10 | 0.0415 (7) | 0.0391 (6) | 0.0425 (7) | 0.0025 (5) | 0.0163 (5) | 0.0012 (5) |
| C11 | 0.0409 (7) | 0.0379 (6) | 0.0478 (7) | 0.0011 (5) | 0.0183 (6) | -0.0006 (5) |
| C12 | 0.0453 (8) | 0.0456 (7) | 0.0636 (9) | -0.0032 (6) | 0.0156 (7) | -0.0091 (7) |

| | | | | | | |
|-----|-------------|------------|-------------|-------------|------------|-------------|
| C13 | 0.0622 (9) | 0.0455 (8) | 0.0635 (10) | -0.0025 (7) | 0.0235 (8) | -0.0146 (7) |
| C14 | 0.0593 (9) | 0.0464 (8) | 0.0671 (10) | 0.0065 (7) | 0.0316 (8) | -0.0072 (7) |
| C15 | 0.0438 (7) | 0.0482 (8) | 0.0584 (9) | 0.0067 (6) | 0.0204 (6) | -0.0013 (7) |
| C16 | 0.0326 (6) | 0.0402 (6) | 0.0428 (7) | -0.0004 (5) | 0.0153 (5) | 0.0025 (5) |
| C17 | 0.0450 (7) | 0.0404 (7) | 0.0491 (8) | 0.0013 (6) | 0.0074 (6) | 0.0064 (6) |
| C18 | 0.0491 (8) | 0.0343 (6) | 0.0588 (9) | 0.0006 (6) | 0.0122 (6) | 0.0003 (6) |
| C19 | 0.0354 (6) | 0.0454 (7) | 0.0443 (7) | -0.0029 (5) | 0.0138 (5) | -0.0025 (6) |
| C20 | 0.0443 (7) | 0.0460 (7) | 0.0448 (7) | 0.0063 (6) | 0.0116 (6) | 0.0055 (6) |
| C21 | 0.0412 (7) | 0.0382 (6) | 0.0457 (7) | 0.0041 (5) | 0.0153 (6) | 0.0028 (6) |
| C22 | 0.0650 (10) | 0.0555 (9) | 0.0642 (10) | 0.0023 (8) | 0.0194 (8) | -0.0144 (8) |

Geometric parameters (Å, °)

| | | | |
|-----------|-------------|--------------|-------------|
| C1—O1 | 1.3467 (17) | C12—C13 | 1.372 (2) |
| C1—C2 | 1.3931 (19) | C12—H12A | 0.9300 |
| C1—C6 | 1.4188 (17) | C13—C14 | 1.388 (2) |
| C2—C3 | 1.374 (2) | C13—H13A | 0.9300 |
| C2—H2A | 0.9300 | C14—C15 | 1.374 (2) |
| C3—C4 | 1.388 (2) | C14—H14A | 0.9300 |
| C3—H3A | 0.9300 | C15—H15A | 0.9300 |
| C4—C5 | 1.3791 (19) | C16—C17 | 1.3928 (18) |
| C4—H4A | 0.9300 | C16—C21 | 1.4028 (18) |
| C5—C6 | 1.3970 (18) | C17—C18 | 1.388 (2) |
| C5—H5A | 0.9300 | C17—H17A | 0.9300 |
| C6—C7 | 1.4718 (16) | C18—C19 | 1.3821 (19) |
| C7—N2 | 1.2924 (16) | C18—H18A | 0.9300 |
| C7—C8 | 1.5101 (17) | C19—O2 | 1.3707 (16) |
| C8—C9 | 1.5129 (18) | C19—C20 | 1.3907 (19) |
| C8—H8A | 0.9700 | C20—C21 | 1.3684 (19) |
| C8—H8B | 0.9700 | C20—H20A | 0.9300 |
| C9—N1 | 1.2927 (16) | C21—H21A | 0.9300 |
| C9—C16 | 1.4751 (18) | C22—O2 | 1.4135 (18) |
| C10—N1 | 1.4001 (17) | C22—H22A | 0.9600 |
| C10—C15 | 1.4095 (18) | C22—H22B | 0.9600 |
| C10—C11 | 1.4146 (19) | C22—H22C | 0.9600 |
| C11—C12 | 1.4003 (19) | O1—H1 | 0.92 (2) |
| C11—N2 | 1.4053 (16) | | |
| O1—C1—C2 | 117.71 (12) | C11—C12—H12A | 119.1 |
| O1—C1—C6 | 122.46 (11) | C12—C13—C14 | 119.56 (14) |
| C2—C1—C6 | 119.83 (13) | C12—C13—H13A | 120.2 |
| C3—C2—C1 | 120.77 (13) | C14—C13—H13A | 120.2 |
| C3—C2—H2A | 119.6 | C15—C14—C13 | 119.96 (13) |
| C1—C2—H2A | 119.6 | C15—C14—H14A | 120.0 |
| C2—C3—C4 | 120.44 (13) | C13—C14—H14A | 120.0 |
| C2—C3—H3A | 119.8 | C14—C15—C10 | 121.72 (14) |
| C4—C3—H3A | 119.8 | C14—C15—H15A | 119.1 |
| C5—C4—C3 | 119.13 (14) | C10—C15—H15A | 119.1 |
| C5—C4—H4A | 120.4 | C17—C16—C21 | 117.23 (12) |
| C3—C4—H4A | 120.4 | C17—C16—C9 | 123.14 (12) |

supplementary materials

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|-----------------|--------------|-----------------|--------------|
| C4—C5—C6 | 122.33 (13) | C21—C16—C9 | 119.63 (11) |
| C4—C5—H5A | 118.8 | C18—C17—C16 | 121.75 (13) |
| C6—C5—H5A | 118.8 | C18—C17—H17A | 119.1 |
| C5—C6—C1 | 117.46 (11) | C16—C17—H17A | 119.1 |
| C5—C6—C7 | 121.95 (11) | C19—C18—C17 | 119.37 (13) |
| C1—C6—C7 | 120.54 (11) | C19—C18—H18A | 120.3 |
| N2—C7—C6 | 118.71 (11) | C17—C18—H18A | 120.3 |
| N2—C7—C8 | 119.07 (11) | O2—C19—C18 | 124.97 (12) |
| C6—C7—C8 | 122.19 (11) | O2—C19—C20 | 115.11 (12) |
| C7—C8—C9 | 105.54 (10) | C18—C19—C20 | 119.91 (13) |
| C7—C8—H8A | 110.6 | C21—C20—C19 | 120.07 (13) |
| C9—C8—H8A | 110.6 | C21—C20—H20A | 120.0 |
| C7—C8—H8B | 110.6 | C19—C20—H20A | 120.0 |
| C9—C8—H8B | 110.6 | C20—C21—C16 | 121.54 (12) |
| H8A—C8—H8B | 108.8 | C20—C21—H21A | 119.2 |
| N1—C9—C16 | 118.30 (12) | C16—C21—H21A | 119.2 |
| N1—C9—C8 | 120.58 (12) | O2—C22—H22A | 109.5 |
| C16—C9—C8 | 121.08 (11) | O2—C22—H22B | 109.5 |
| N1—C10—C15 | 116.04 (12) | H22A—C22—H22B | 109.5 |
| N1—C10—C11 | 125.82 (11) | O2—C22—H22C | 109.5 |
| C15—C10—C11 | 117.85 (12) | H22A—C22—H22C | 109.5 |
| C12—C11—N2 | 116.55 (12) | H22B—C22—H22C | 109.5 |
| C12—C11—C10 | 118.96 (12) | C9—N1—C10 | 121.56 (11) |
| N2—C11—C10 | 124.11 (12) | C7—N2—C11 | 122.42 (11) |
| C13—C12—C11 | 121.76 (14) | C1—O1—H1 | 105.1 (13) |
| C13—C12—H12A | 119.1 | C19—O2—C22 | 118.25 (12) |
| O1—C1—C2—C3 | 178.92 (14) | C13—C14—C15—C10 | -2.2 (2) |
| C6—C1—C2—C3 | -1.7 (2) | N1—C10—C15—C14 | 178.36 (13) |
| C1—C2—C3—C4 | 1.3 (2) | C11—C10—C15—C14 | 4.2 (2) |
| C2—C3—C4—C5 | 0.4 (2) | N1—C9—C16—C17 | 163.12 (12) |
| C3—C4—C5—C6 | -1.8 (2) | C8—C9—C16—C17 | -19.07 (18) |
| C4—C5—C6—C1 | 1.4 (2) | N1—C9—C16—C21 | -16.64 (17) |
| C4—C5—C6—C7 | -176.28 (13) | C8—C9—C16—C21 | 161.17 (11) |
| O1—C1—C6—C5 | 179.68 (13) | C21—C16—C17—C18 | -2.89 (19) |
| C2—C1—C6—C5 | 0.36 (19) | C9—C16—C17—C18 | 177.35 (12) |
| O1—C1—C6—C7 | -2.6 (2) | C16—C17—C18—C19 | 0.6 (2) |
| C2—C1—C6—C7 | 178.09 (12) | C17—C18—C19—O2 | -177.80 (12) |
| C5—C6—C7—N2 | 174.04 (12) | C17—C18—C19—C20 | 2.7 (2) |
| C1—C6—C7—N2 | -3.59 (18) | O2—C19—C20—C21 | 176.78 (12) |
| C5—C6—C7—C8 | -3.92 (19) | C18—C19—C20—C21 | -3.64 (19) |
| C1—C6—C7—C8 | 178.45 (11) | C19—C20—C21—C16 | 1.3 (2) |
| N2—C7—C8—C9 | -73.70 (14) | C17—C16—C21—C20 | 1.90 (19) |
| C6—C7—C8—C9 | 104.25 (13) | C9—C16—C21—C20 | -178.32 (11) |
| C7—C8—C9—N1 | 71.22 (14) | C16—C9—N1—C10 | 174.75 (11) |
| C7—C8—C9—C16 | -106.54 (12) | C8—C9—N1—C10 | -3.08 (18) |
| N1—C10—C11—C12 | -175.71 (13) | C15—C10—N1—C9 | 148.84 (12) |
| C15—C10—C11—C12 | -2.14 (19) | C11—C10—N1—C9 | -37.48 (19) |
| N1—C10—C11—N2 | -3.1 (2) | C6—C7—N2—C11 | -171.94 (11) |
| C15—C10—C11—N2 | 170.50 (12) | C8—C7—N2—C11 | 6.09 (19) |

| | | | |
|-----------------|--------------|----------------|--------------|
| N2—C11—C12—C13 | -174.98 (13) | C12—C11—N2—C7 | -147.30 (13) |
| C10—C11—C12—C13 | -1.8 (2) | C10—C11—N2—C7 | 39.91 (19) |
| C11—C12—C13—C14 | 3.8 (2) | C18—C19—O2—C22 | 12.9 (2) |
| C12—C13—C14—C15 | -1.8 (2) | C20—C19—O2—C22 | -167.51 (13) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|-----------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| O1—H1 \cdots N2 | 0.92 (2) | 1.71 (2) | 2.5563 (17) | 151.5 (19) |
| C14—H14A \cdots O2 ⁱ | 0.93 | 2.53 | 3.406 (2) | 156 |

Symmetry codes: (i) $-x+3/2, y-1/2, -z+3/2$.

Fig. 1

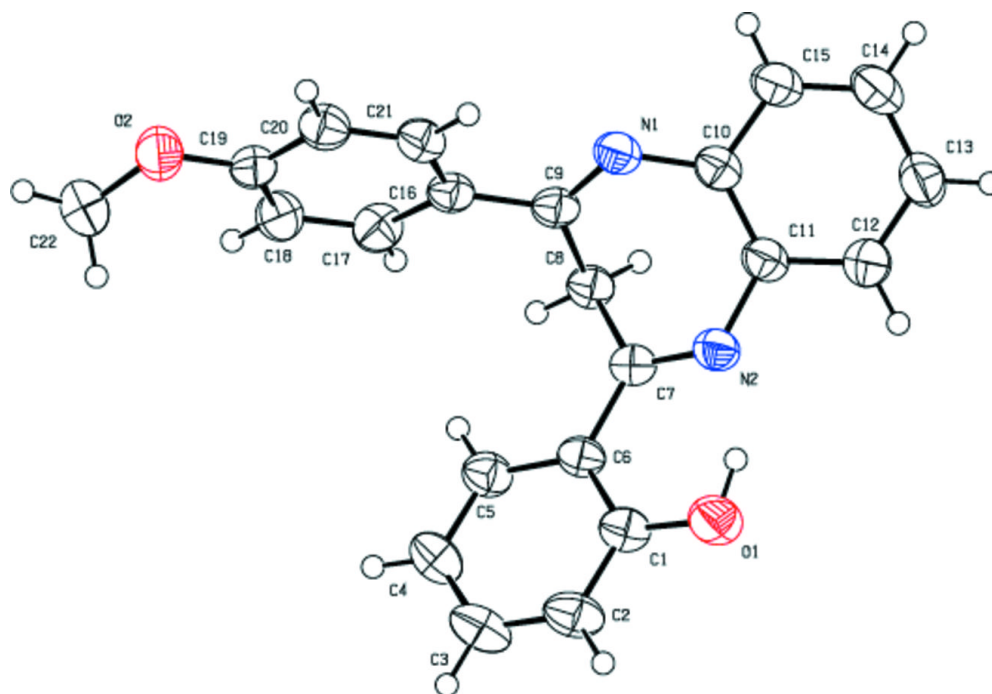


Fig. 2

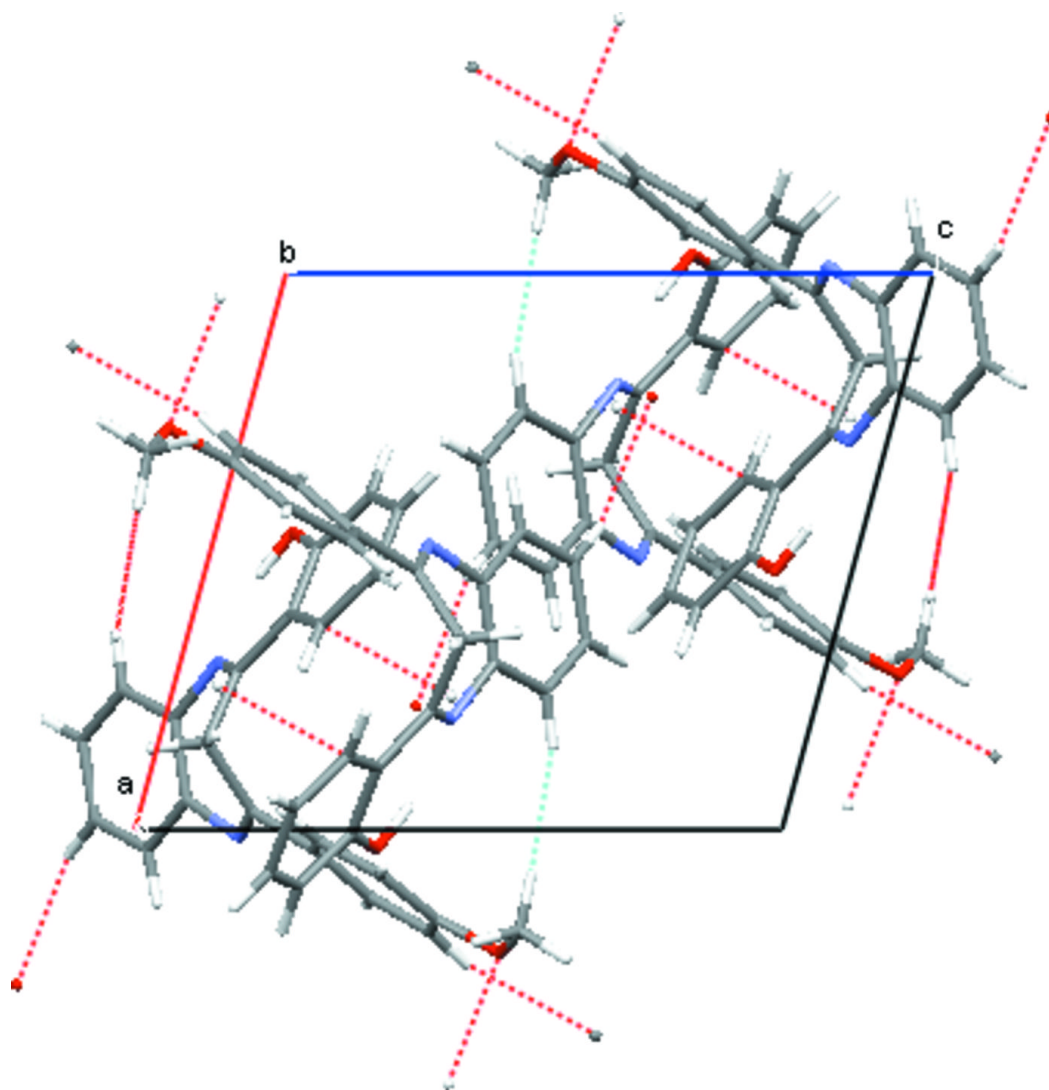


Fig. 3

